

[2-(*N,N*-Dimethylamino)-5-methylphenyl]-
diphenylmethanolLouise B. Kranholm, Andrew D.
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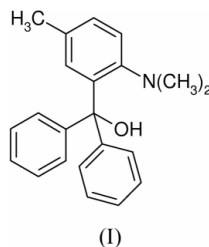
Key indicators

Single-crystal X-ray study
 $T = 180$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.048
 wR factor = 0.153
Data-to-parameter ratio = 19.4For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.The crystal structure of (2-(*N,N*-dimethylamino)-5-methylphenyl)diphenylmethanol, $\text{C}_{22}\text{H}_{23}\text{NO}$, (I), at 180 K contains an intramolecular $\text{N}-\text{H} \cdots \text{O}$ hydrogen bond.

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Experimental

The compound was prepared according to a literature procedure (Ludt *et al.*, 1970). Crystals suitable for X-ray analysis were prepared by evaporation of an ethyl acetate/hexane solution.

Crystal data

$\text{C}_{22}\text{H}_{23}\text{NO}$	$D_x = 1.185$ Mg m $^{-3}$
$M_r = 317.41$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 3897 reflections
$a = 8.9287$ (4) Å	$\theta = 2.7$ – 28.1°
$b = 13.2618$ (6) Å	$\mu = 0.07$ mm $^{-1}$
$c = 15.1880$ (8) Å	$T = 180$ (2) K
$\beta = 98.505$ (1) $^\circ$	Block, colourless
$V = 1778.64$ (15) Å 3	$0.40 \times 0.32 \times 0.24$ mm
$Z = 4$	

Data collection

Bruker–Nonius X8APEXII CCD diffractometer	4347 independent reflections
Thin-slice ω and φ scans	3398 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 2003)	$R_{\text{int}} = 0.022$
$T_{\text{min}} = 0.820$, $T_{\text{max}} = 0.983$	$\theta_{\text{max}} = 28.3^\circ$
10423 measured reflections	$h = -11 \rightarrow 11$
	$k = -17 \rightarrow 9$
	$l = -19 \rightarrow 20$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0959P)^2 + 0.1677P]$
$R[F^2 > 2\sigma(F^2)] = 0.048$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.153$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.08$	$\Delta\rho_{\text{max}} = 0.40$ e Å $^{-3}$
4347 reflections	$\Delta\rho_{\text{min}} = -0.44$ e Å $^{-3}$
224 parameters	
H atoms treated by a mixture of independent and constrained refinement	

Table 1

Hydrogen-bonding geometry (Å, $^\circ$).

$D-\text{H} \cdots A$	$D-\text{H}$	$\text{H} \cdots A$	$D \cdots A$	$D-\text{H} \cdots A$
$\text{O1}-\text{H1} \cdots \text{N1}$	0.88 (2)	1.87 (2)	2.6683 (14)	150.6 (18)

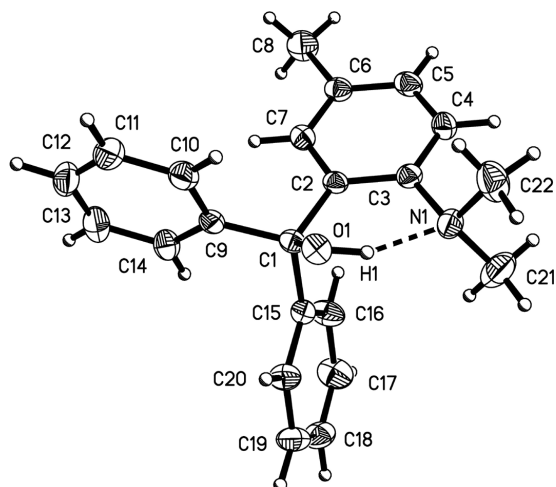


Figure 1

A view of the molecular structure, with displacement ellipsoids drawn at the 50% probability level for non-H atoms. H atoms are shown as spheres of arbitrary radius. The dashed line denotes the intramolecular hydrogen bond.

H atoms bound to C were positioned geometrically and allowed to ride during subsequent refinement: C–H = 0.95 Å, $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$ for H atoms bound to the phenyl rings; C–H = 0.98 Å,

$U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{C})$ for H atoms of the methyl groups. The methyl groups were allowed to rotate about their local threefold axes. H1, associated with the hydroxyl group, was located in a difference Fourier map and refined freely with an isotropic displacement parameter.

Data collection: *APEX2* (Bruker–Nonius, 2003); cell refinement: *SAINT* (Bruker, 2003); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2000); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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